

ELEMENTARY PARTICLE PHYSICS

A BRIEF INTRODUCTION (PART I)

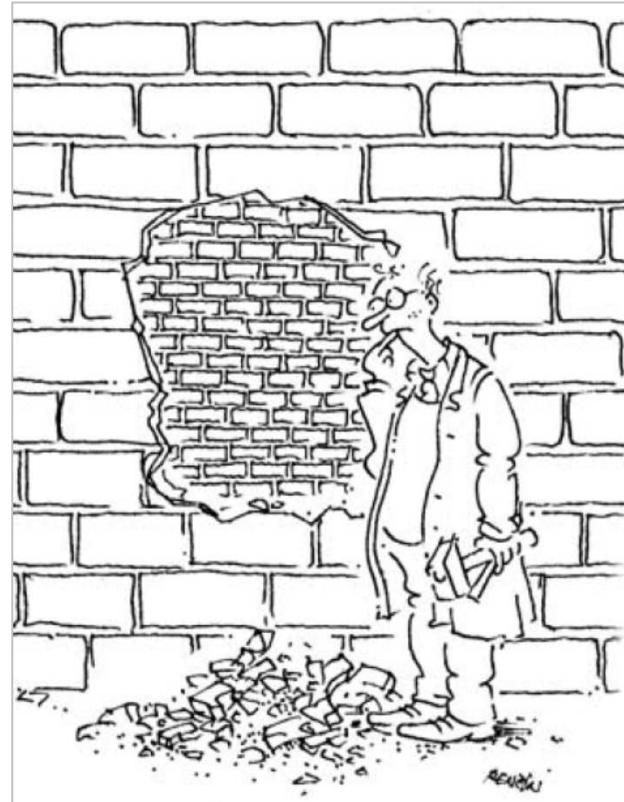
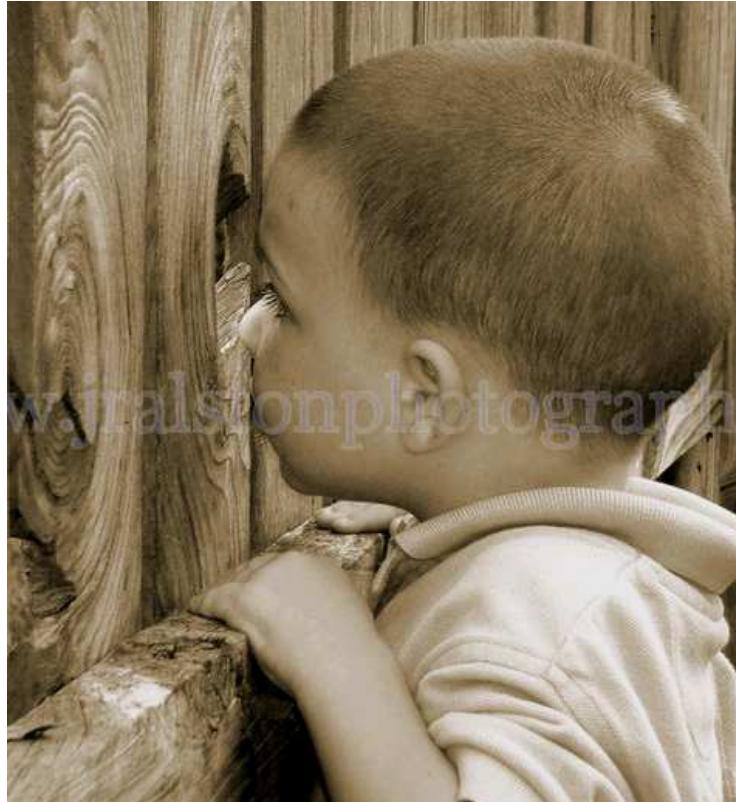
APRIL 2020 | HANS STRÖHER (FZ JÜLICH, UNIVERSITY OF COLOGNE)

Outline – EPP Video Lectures

- An introduction of Elementary Particle Physics (EPP) via video
- Lecturers: Hans Ströher, Irakli Keshelashvili, Detlev Gotta
- Time: Monday & Friday @ 2 pm (Jülich time) → 4 pm (Tbilisi)
- Outline:
 - A brief **introduction** (history ...)
 - The **tools** (accelerators, targets, detectors ... kinematics, ...)
 - The **particles** (hadrons, baryons, mesons)
 - The **fundamental particles** (quarks, leptons)
 - The **forces** (gravitation, nuclear forces)
 - The **fundamental interactions** (strong and electro-weak IA)
 - The **Standard Model** of EPP
 - Physics **Beyond the Standard Model** (BSM)
 - Spin-offs – **Applications** of EPP

Overall Information

Introduction – Prelude



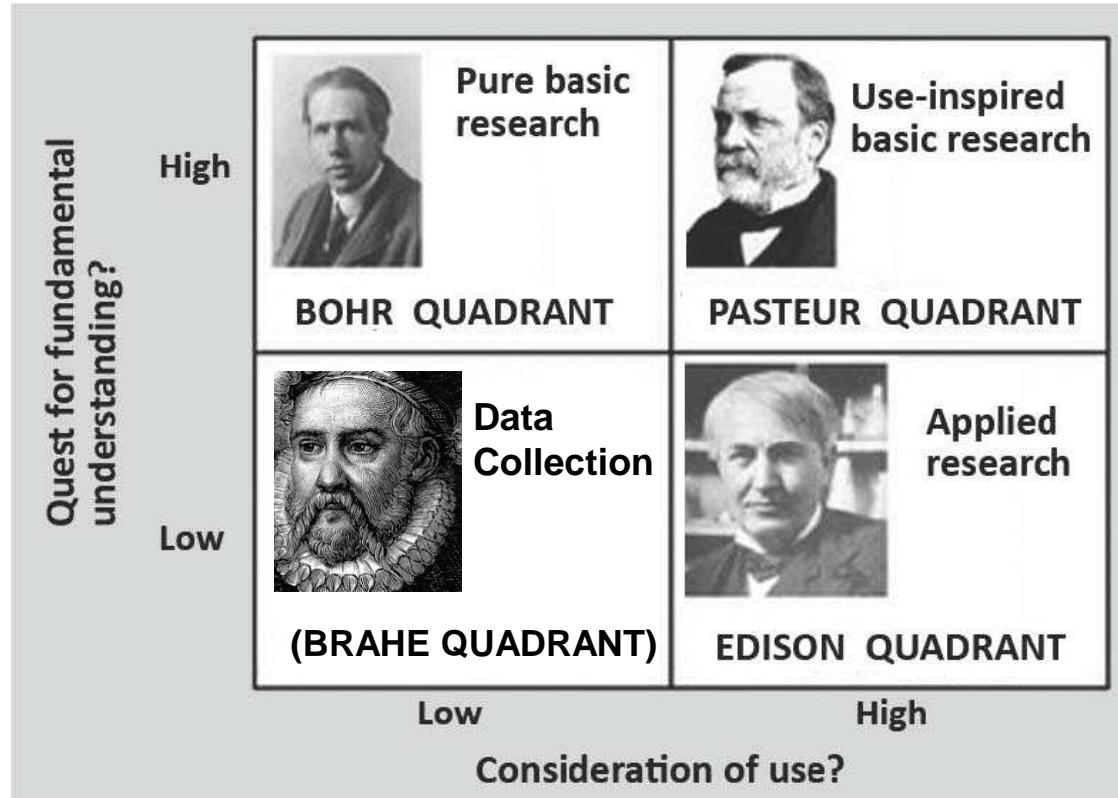
Science is driven by curiosity

Introduction – Prelude

- What we would like to know:
 - What are the **fundamental building blocks** of Nature?
 - Why are there **three generations** of fermions?
 - Are **neutrinos** their own antiparticle?
 - What are the **interactions** between the building blocks?
 - Do all **forces** unify?
 - Where does the **mass of the particles** originate?
 - What is **Dark Matter**? What is the origin of Dark Matter?
 - Or even **Dark Energy**?
 - Why is **antimatter** different from matter?
 - What is the origin of the **matter-antimatter asymmetry**?
 - Is the **proton** (ordinary matter) **stable**?
 - ...

Fundamental questions

Introduction – Prelude



Curiosity-driven vs. use-inspired (basic) research

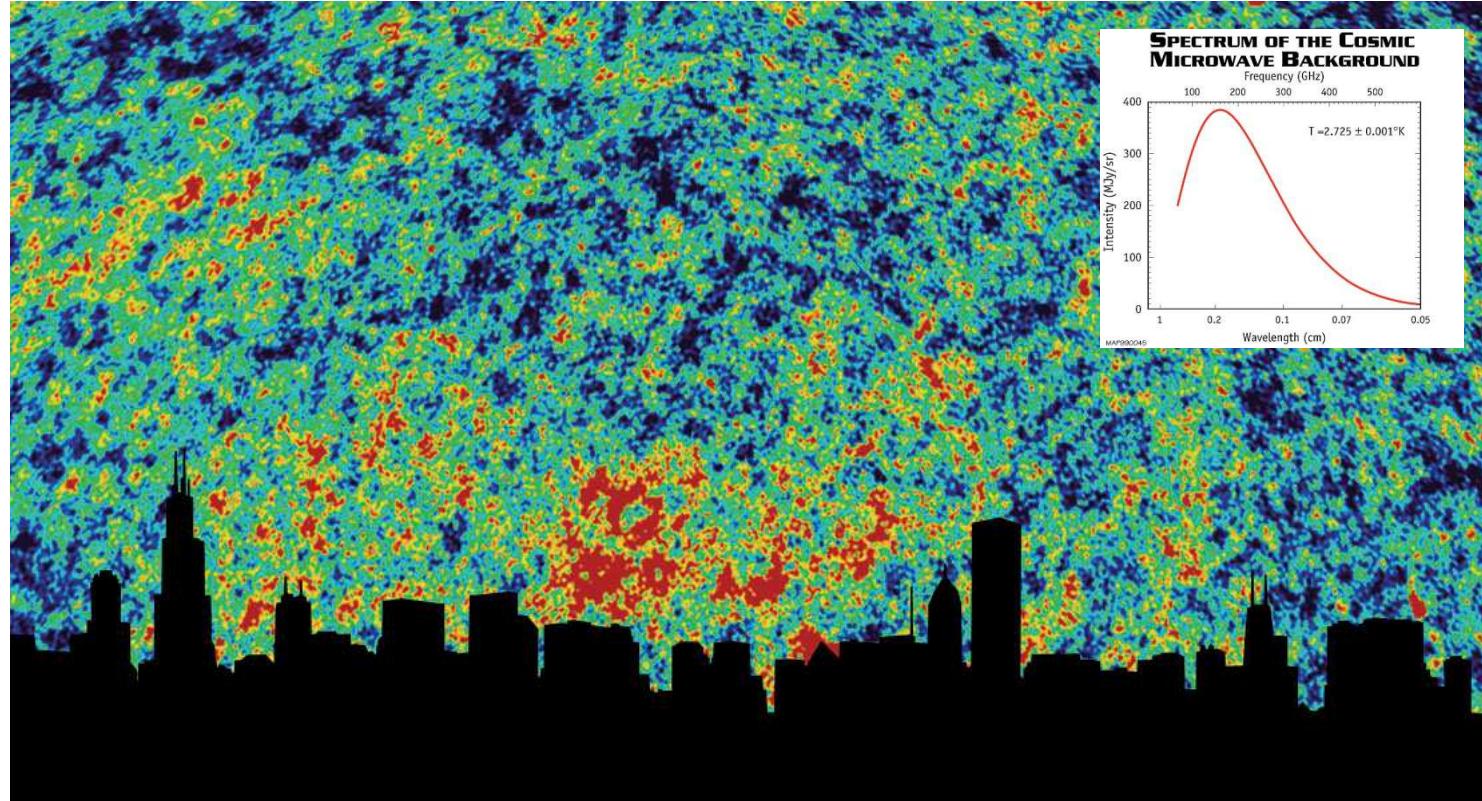
Introduction – Prelude



The *Milky Way* over monument valley

Mankind strives to understand **Nature** – from the very large ...

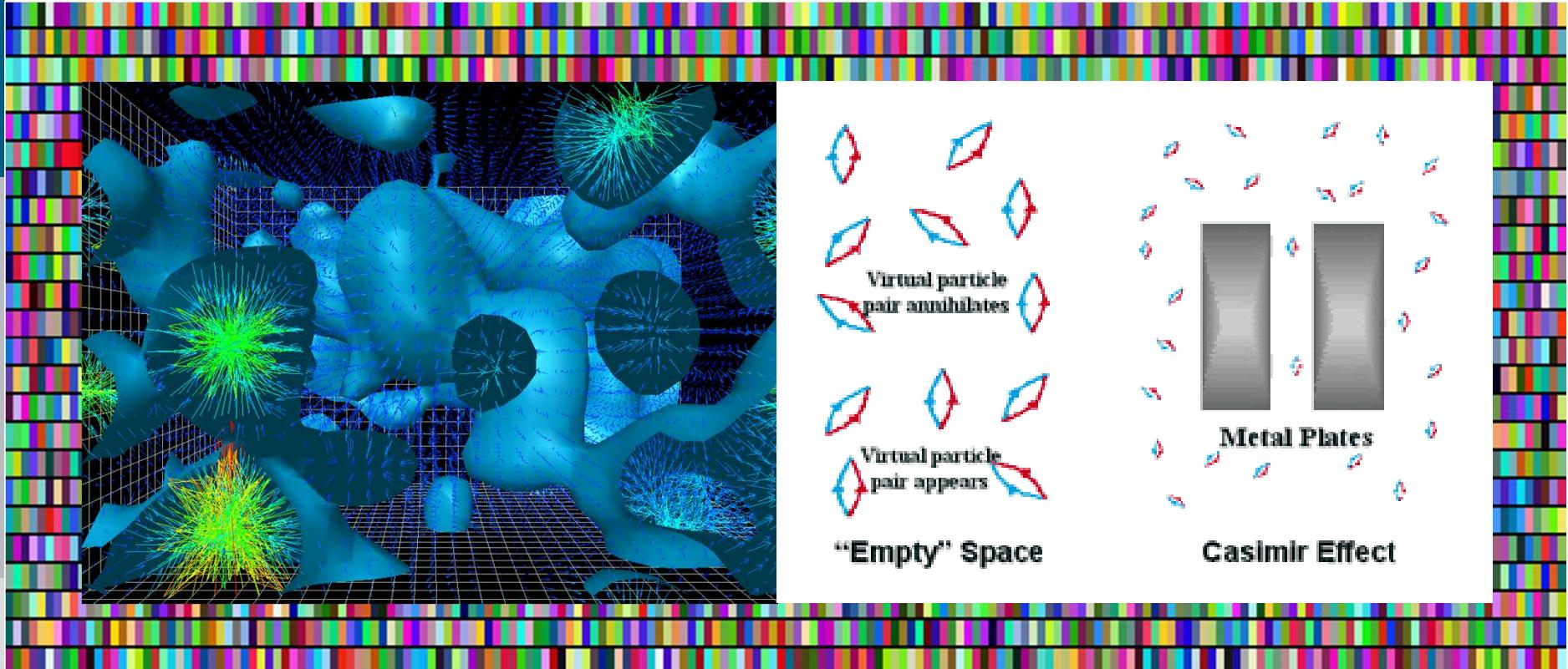
Introduction – Prelude



The Cosmic Microwave Background (CMB)

Mankind strives to understand **Nature** – from the very large ...

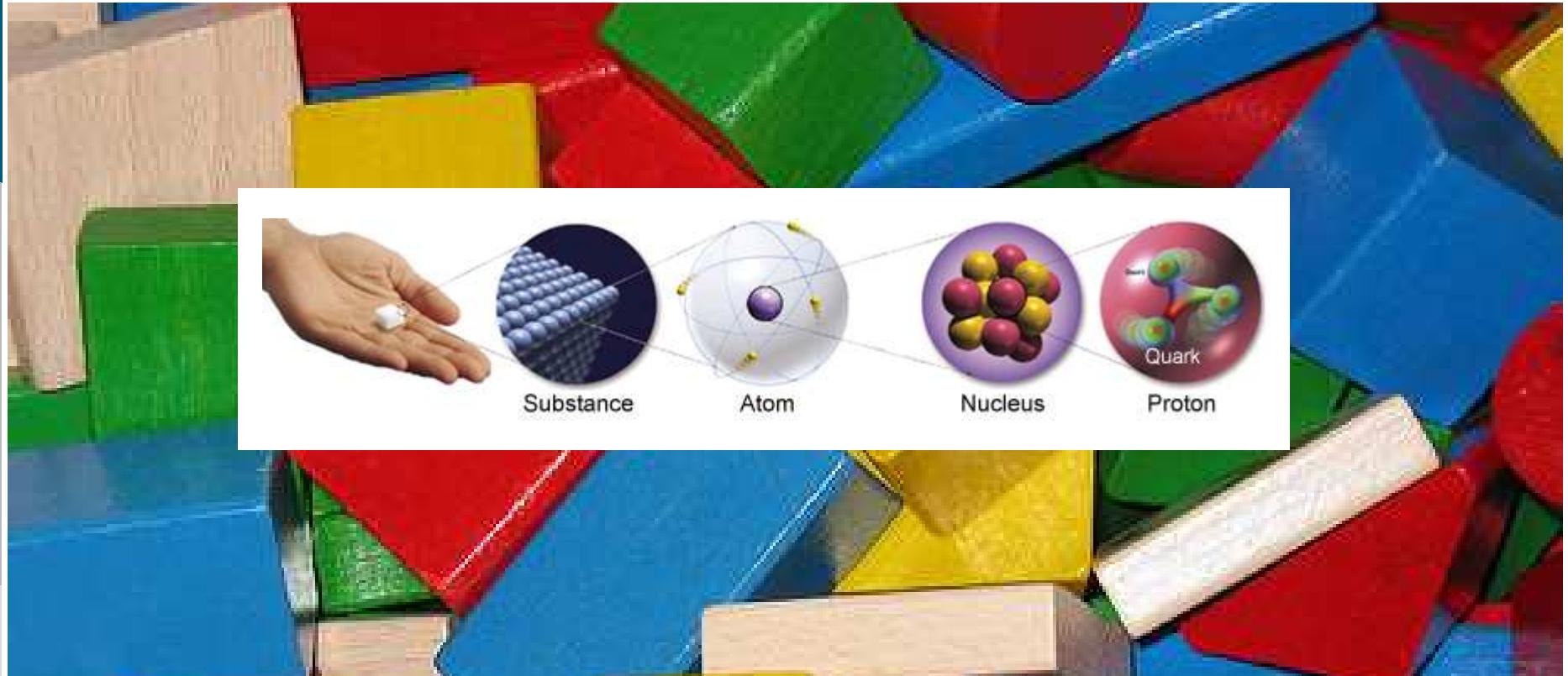
Introduction – Prelude



Vacuum Fluctuations (Simulation)

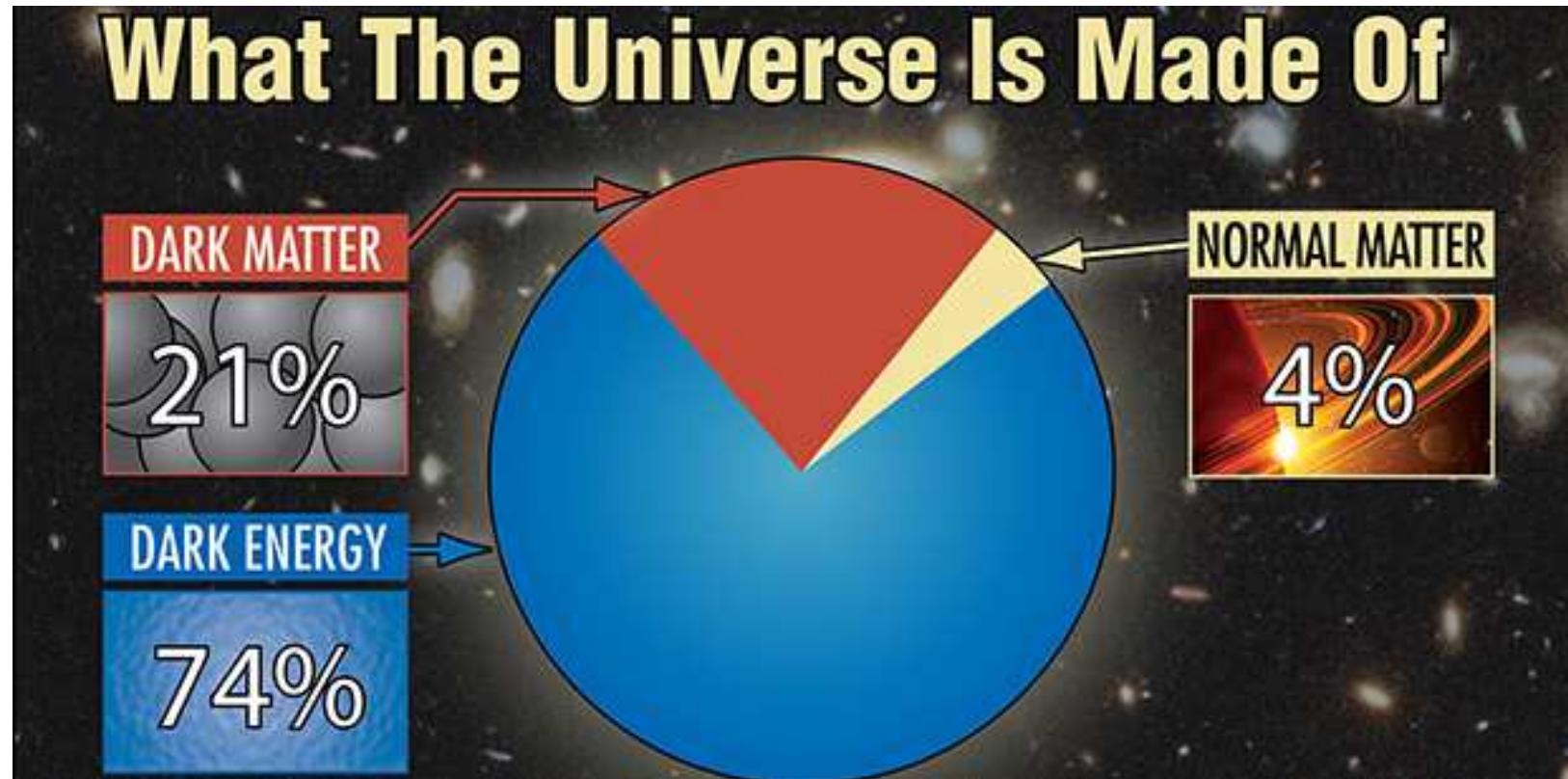
Mankind strives to understand **Nature** – from the very large to the very small

Introduction – Prospect



The building blocks of matter

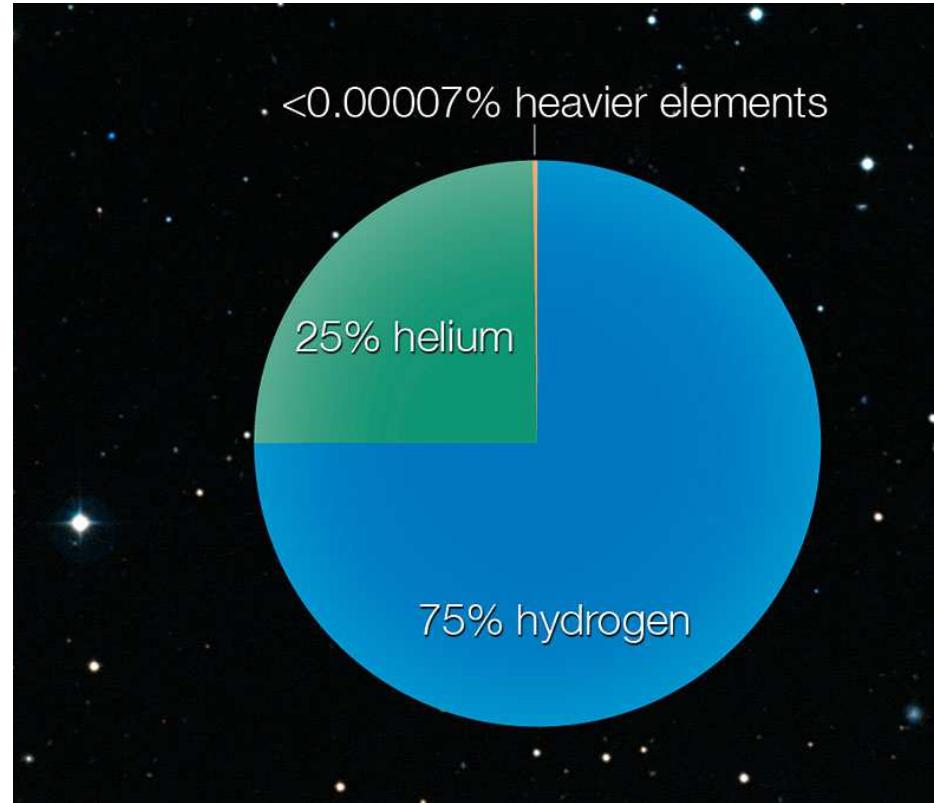
There is a **connection** between the *very large* and the *very small*



What we know and what we don't know

This talk is only about a **tiny fraction** of the energy content of the Universe

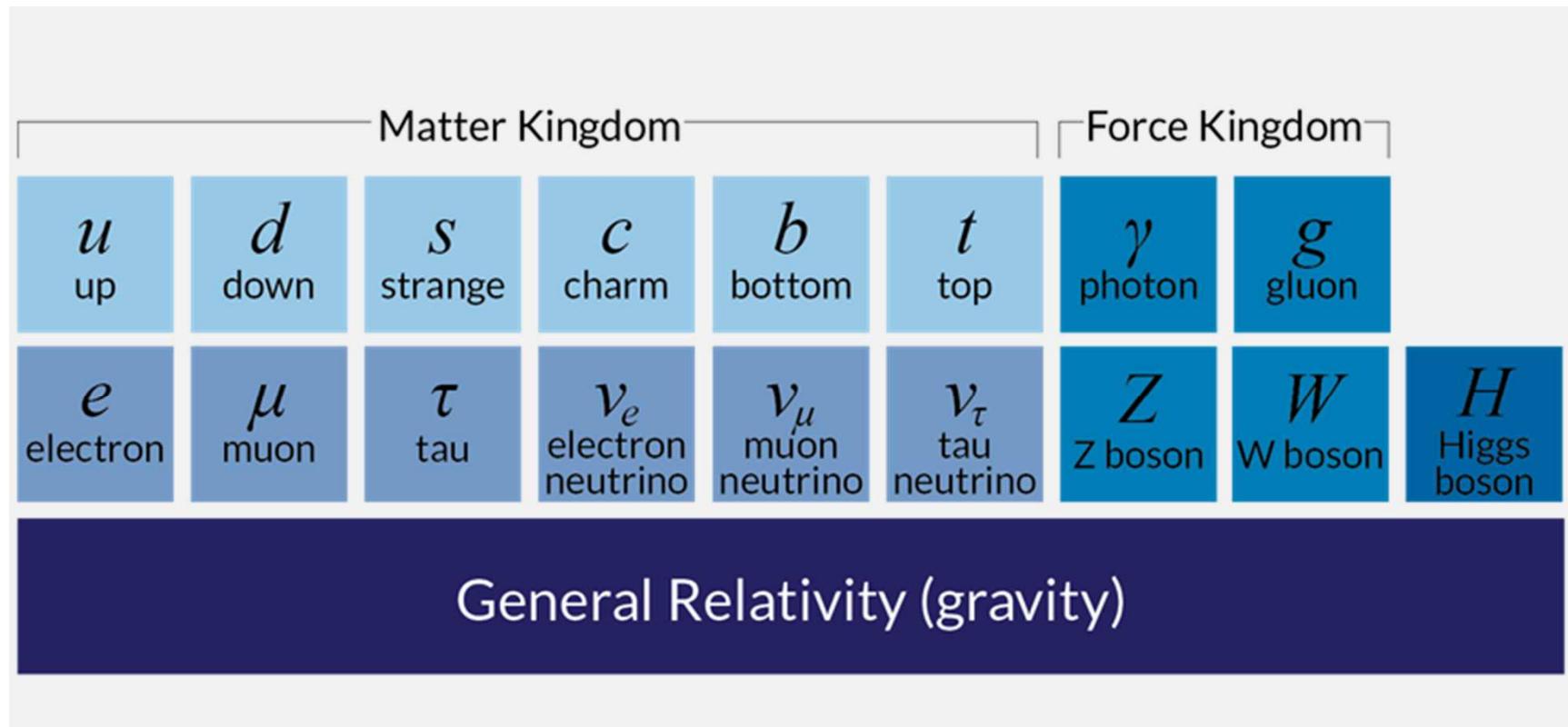
Introduction – Prospect



What we know and what we don't know

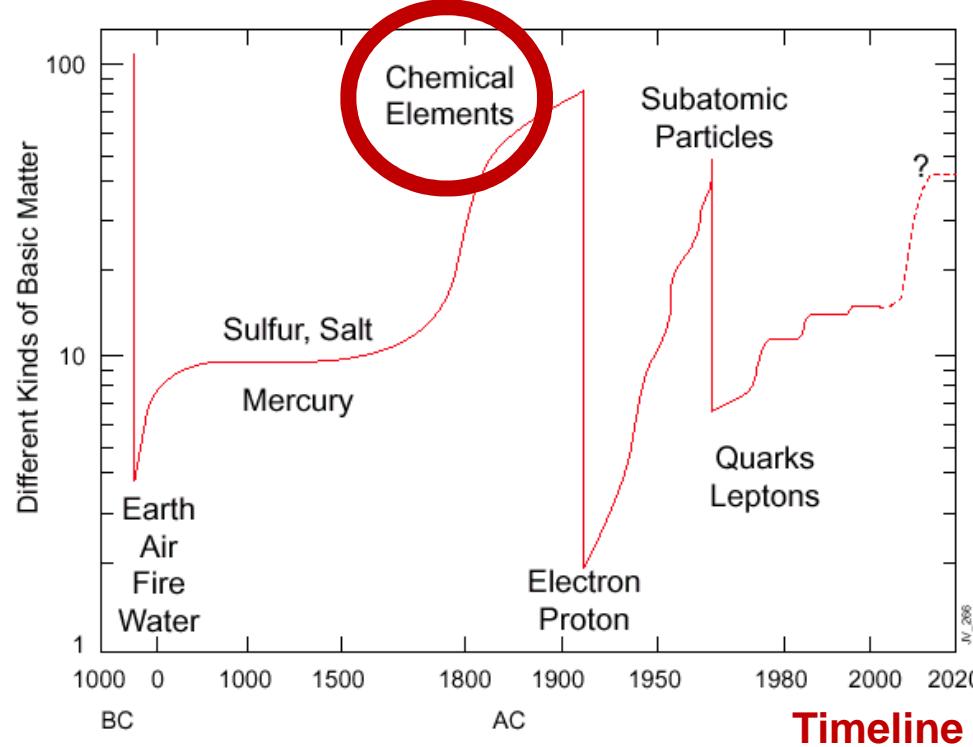
Matter is **essentially hydrogen** and helium (+ a tiny fraction of heavy nuclei)

Introduction – Prospect



F. Wilczek (Nobel Laureate Physics 2004): „Core Theory“ of physics

Introduction – History



A historical account of the number of **matter-constituents**

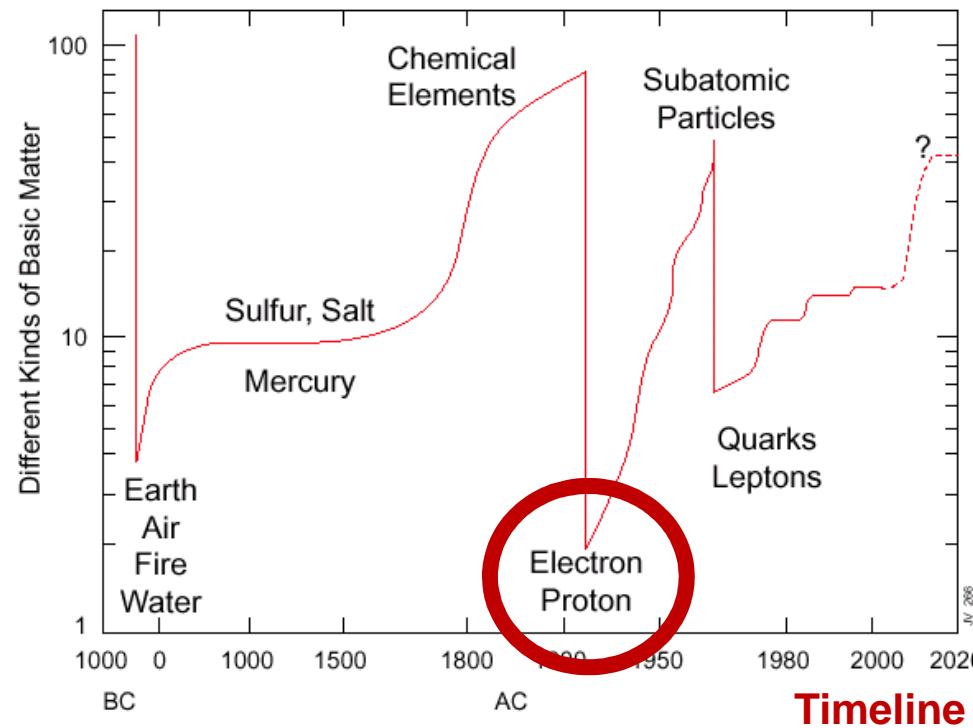
Introduction – History

1 H Hydrogen														2 He Helium				
3 Li Lithium	4 Be Beryllium																	
11 Na Sodium	12 Mg Magnesium																	
19 K Potassium	20 Ca Calcium																	
37 Rb Rubidium	38 Sr Strontium																	
55 Cs Cesium	56 Ba Barium	*	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton
87 Fr Francium	88 Ra Radium	*	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon
		*	71 Lu Lutetium	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon
		*	103 Lr Lawrencium	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Darmstadtium	111 Rg Roentgenium	112 Cn Copernicium		114 Fl Flerovium		116 Lv Livermorium		
		*	57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium		
		*	89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium		

Pauli-principle at work

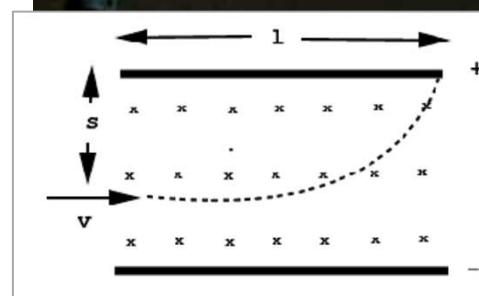
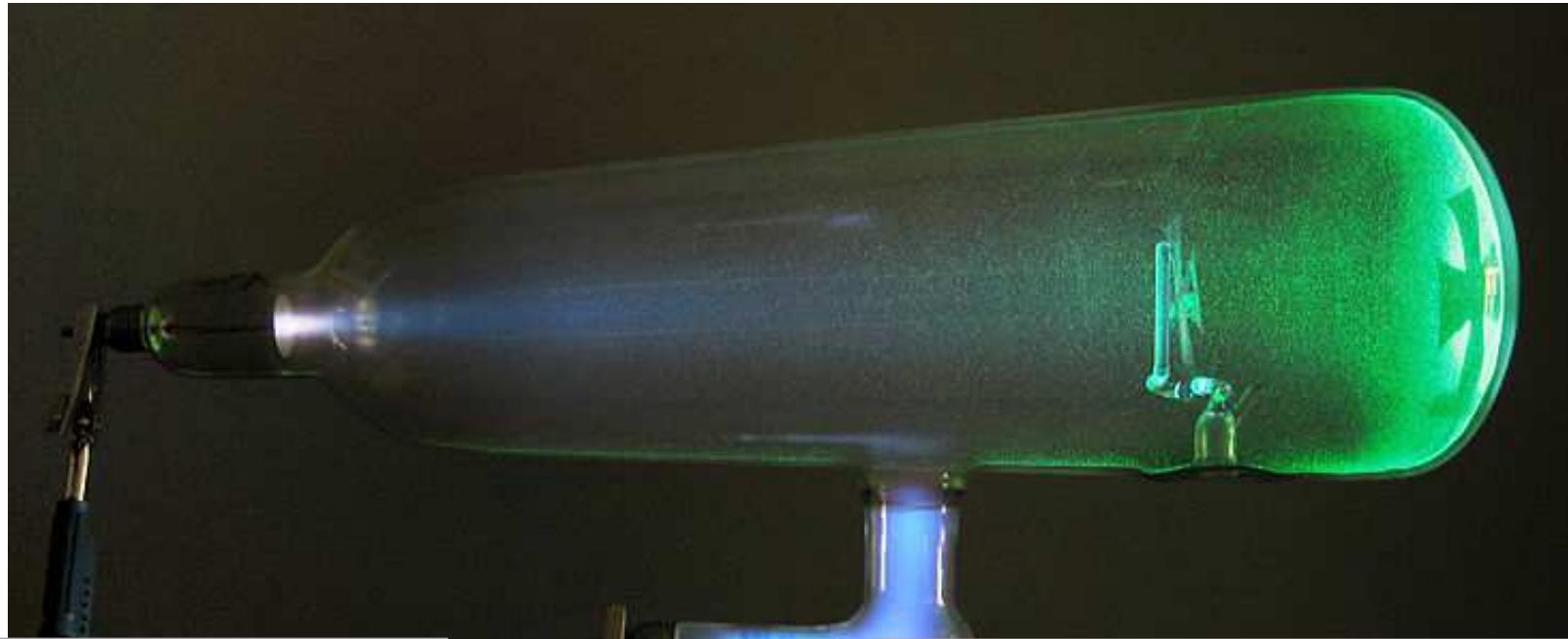
Periodic Table of Elements

Introduction – History



The up's and down's in the number of matter constituents

Introduction – History



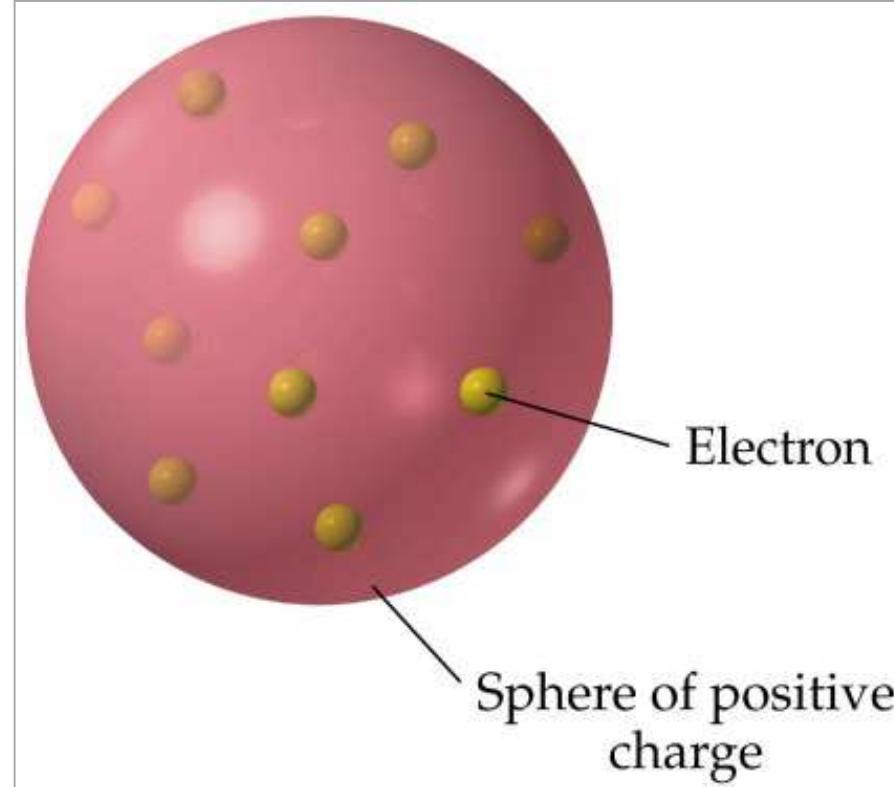
$$\frac{e}{m} = \frac{2sE}{l^2H^2}$$

[The charge-to-mass ratio of cathode rays, e/m]
[The charge-to-mass ratio of hydrogen ion]

$$= \frac{1.76 \times 10^{11}}{9.65 \times 10^7} \\ \approx 1800.$$

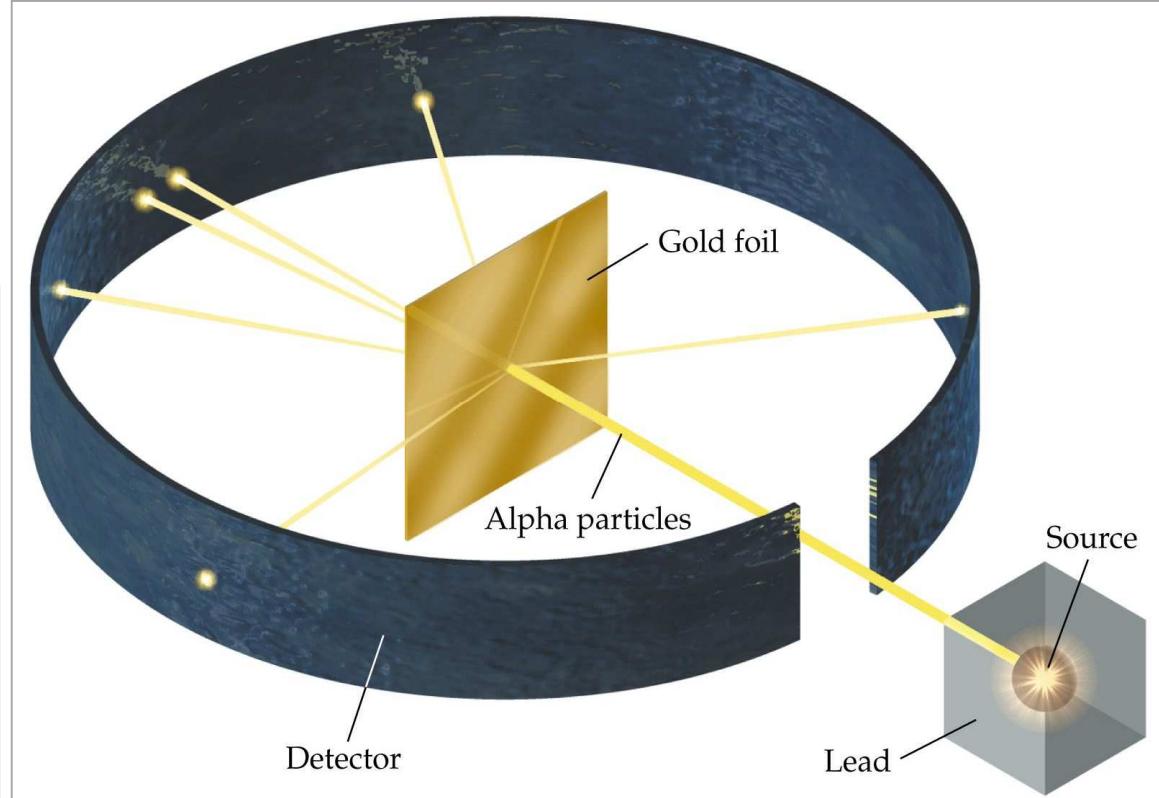
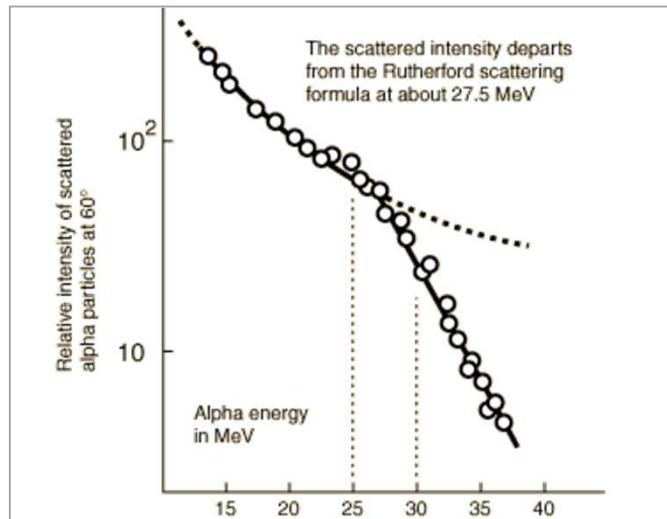
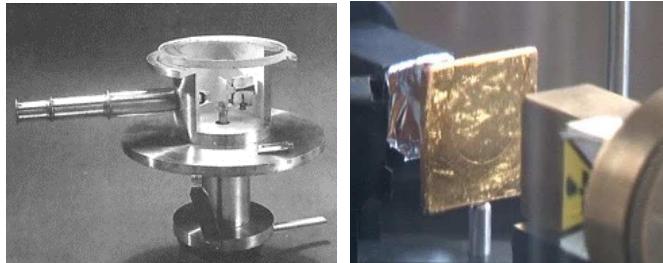
„Cathode rays“ → discovery of the **electron**

Introduction – History



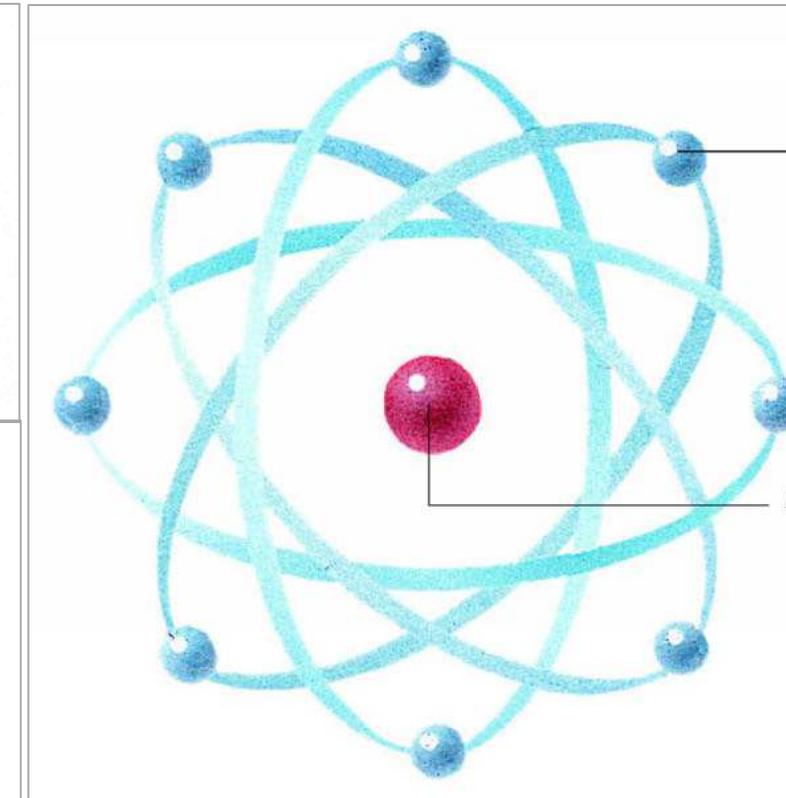
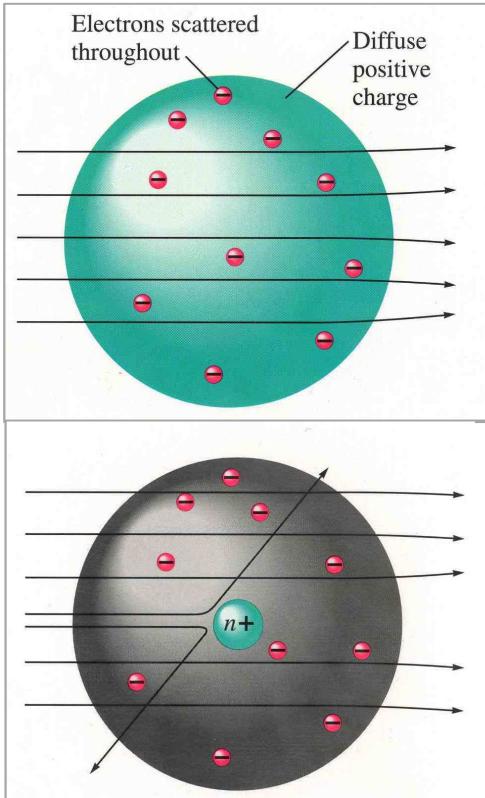
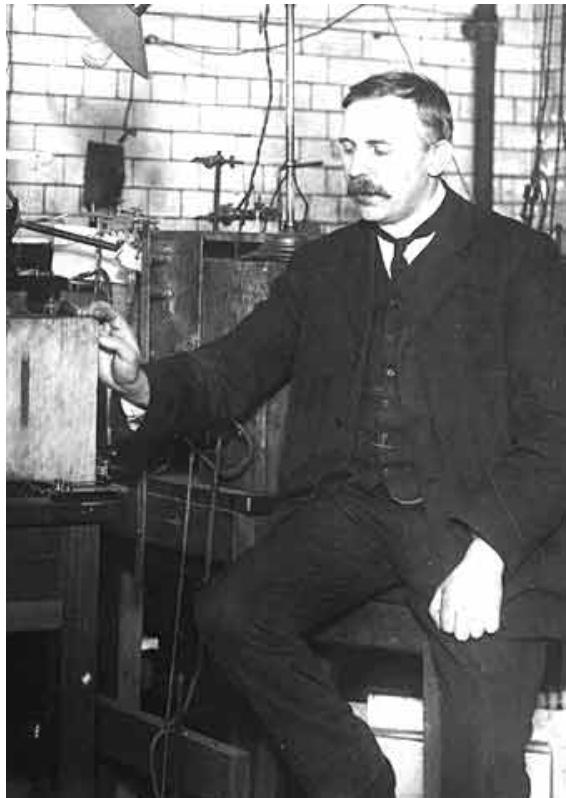
Discovery of the **electron** (J.J. Thomson, 1897) → Thomson's atom-model

Introduction – History



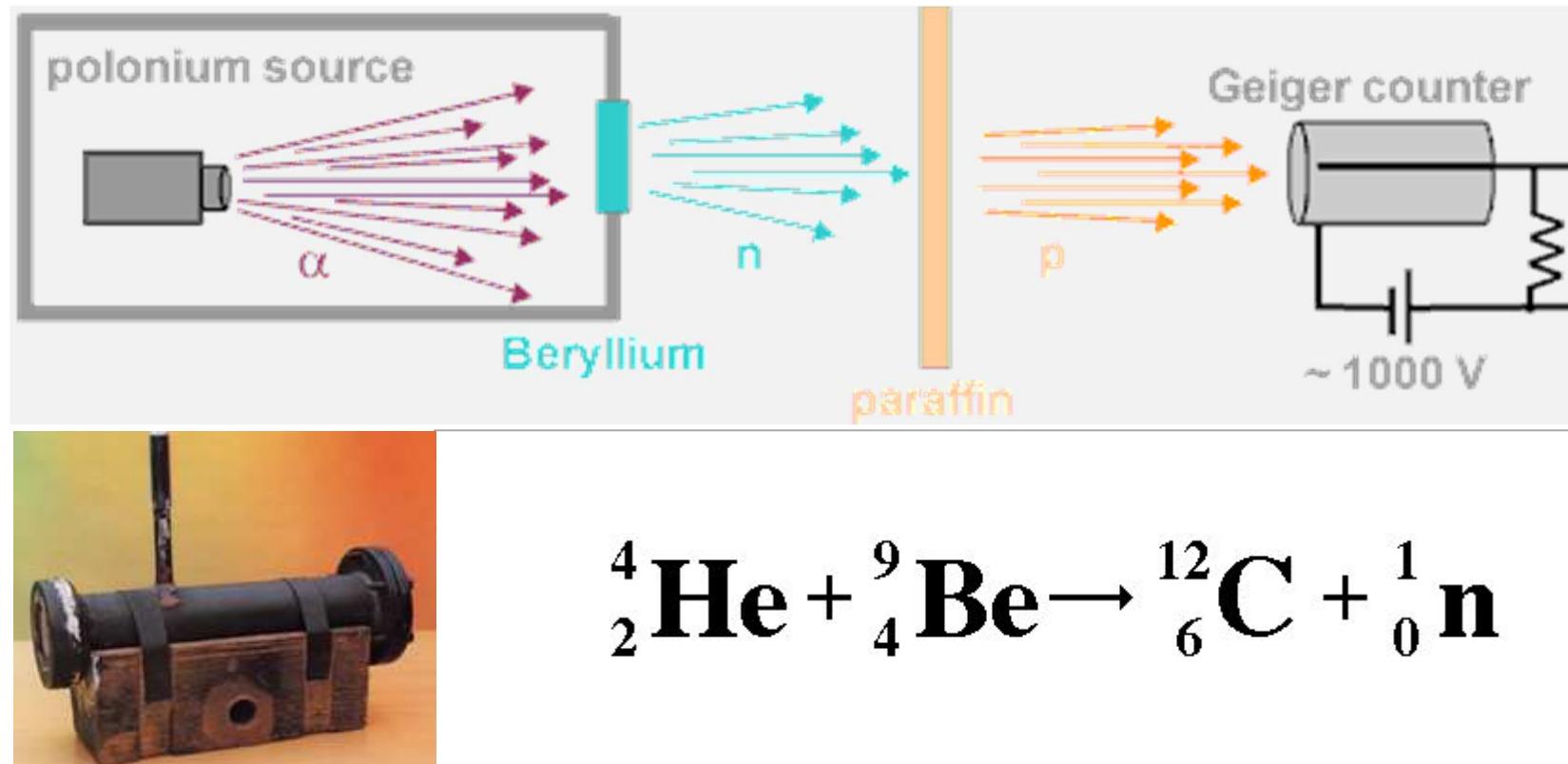
Discovery of the atomic **nucleus** (E. Rutherford, 1911) → new atomic model

Introduction – History



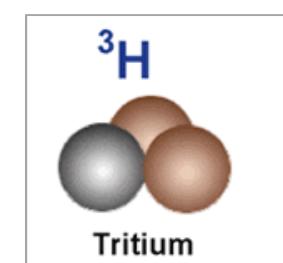
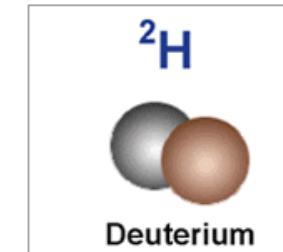
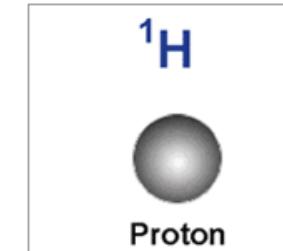
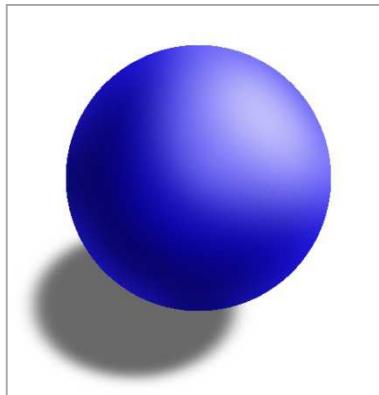
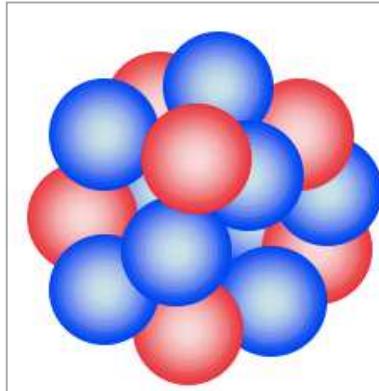
Discovery of the atomic **nucleus** (E. Rutherford, 1911) → new atomic model

Introduction – History



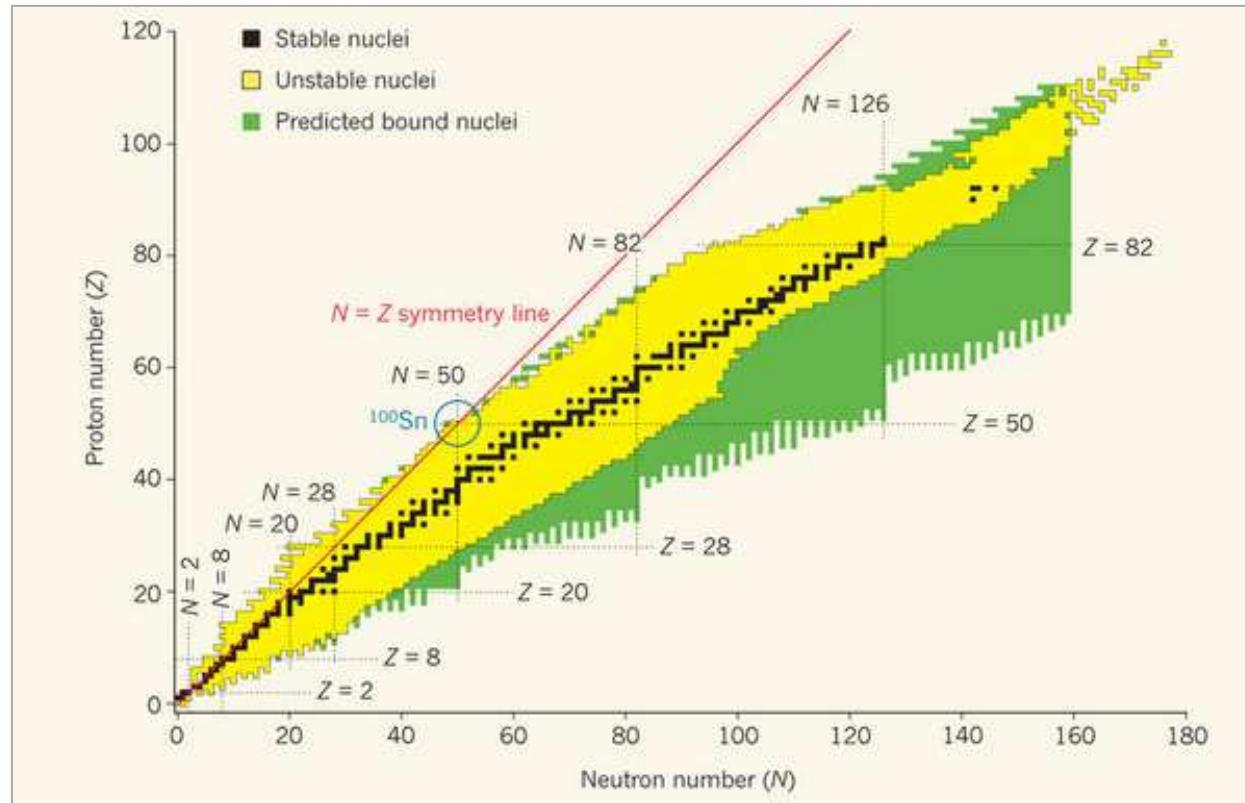
Discovery of the **neutron** (J. Chadwick, 1932) → structure of the nucleus

Introduction – History



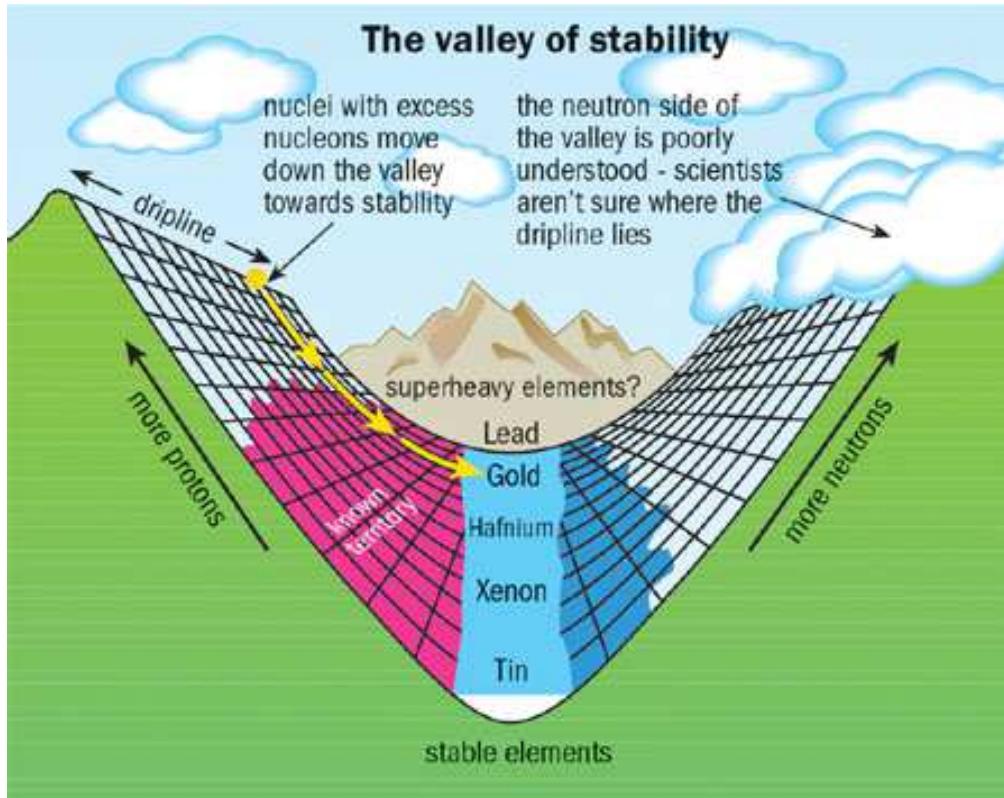
Discovery of the **neutron** (J. Chadwick, 1932) → isotopes

Introduction – History

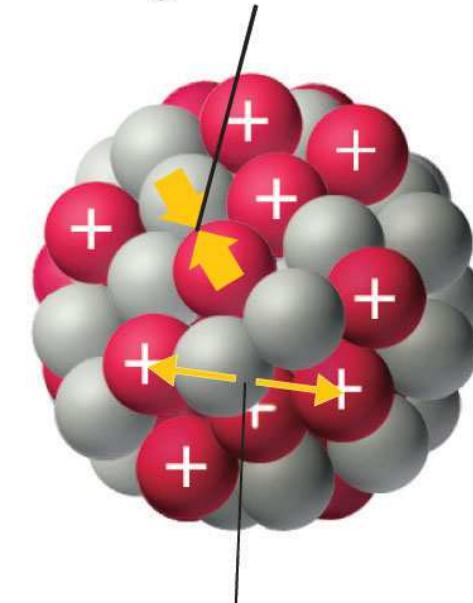


The „Table of Isotopes“

Introduction – History



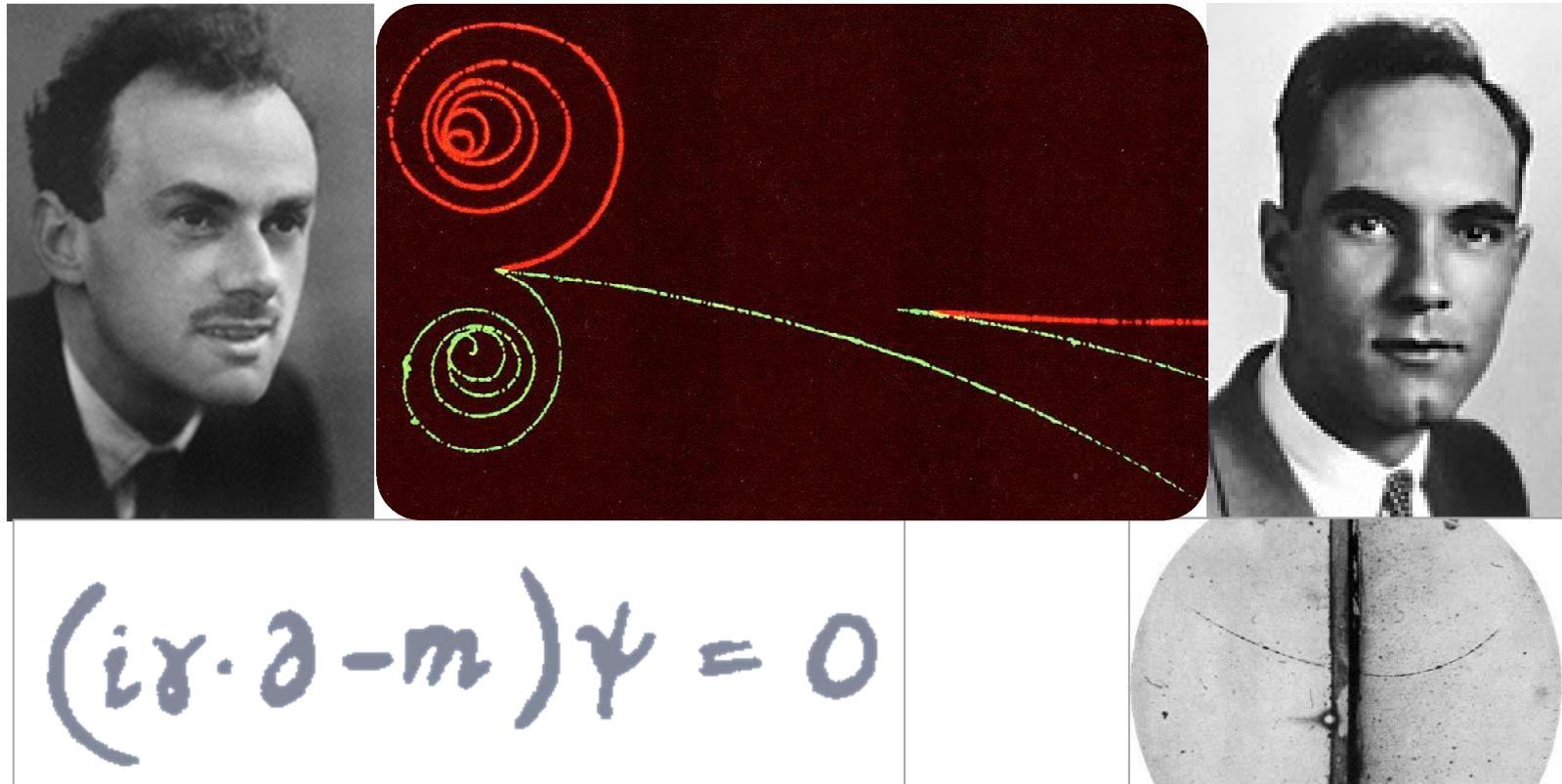
Strong nuclear force



Electrostatic repulsion

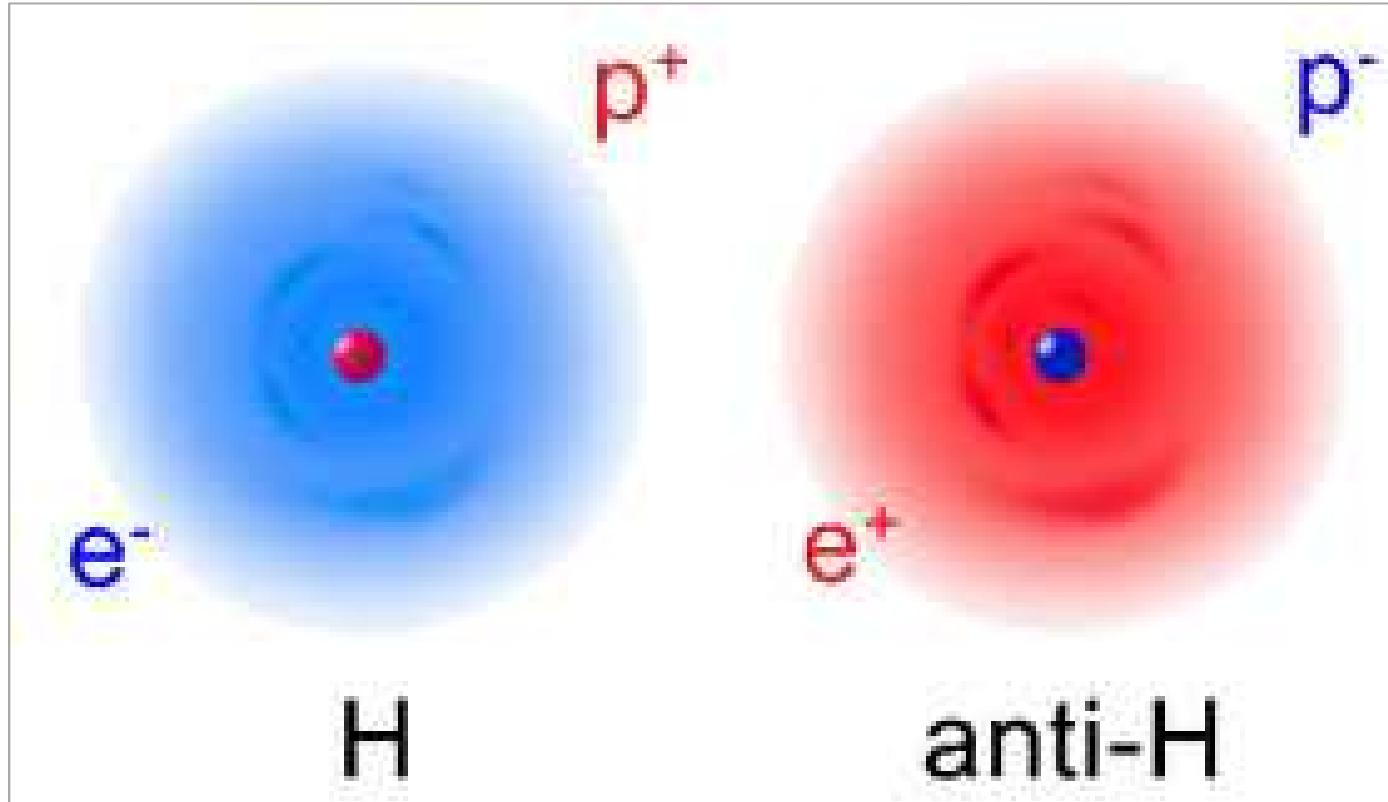
The „Valley of Stability“

Introduction – History



Discovery of **antiparticles** (P.A.M. Dirac, 1928, C.D. Anderson, 1932)

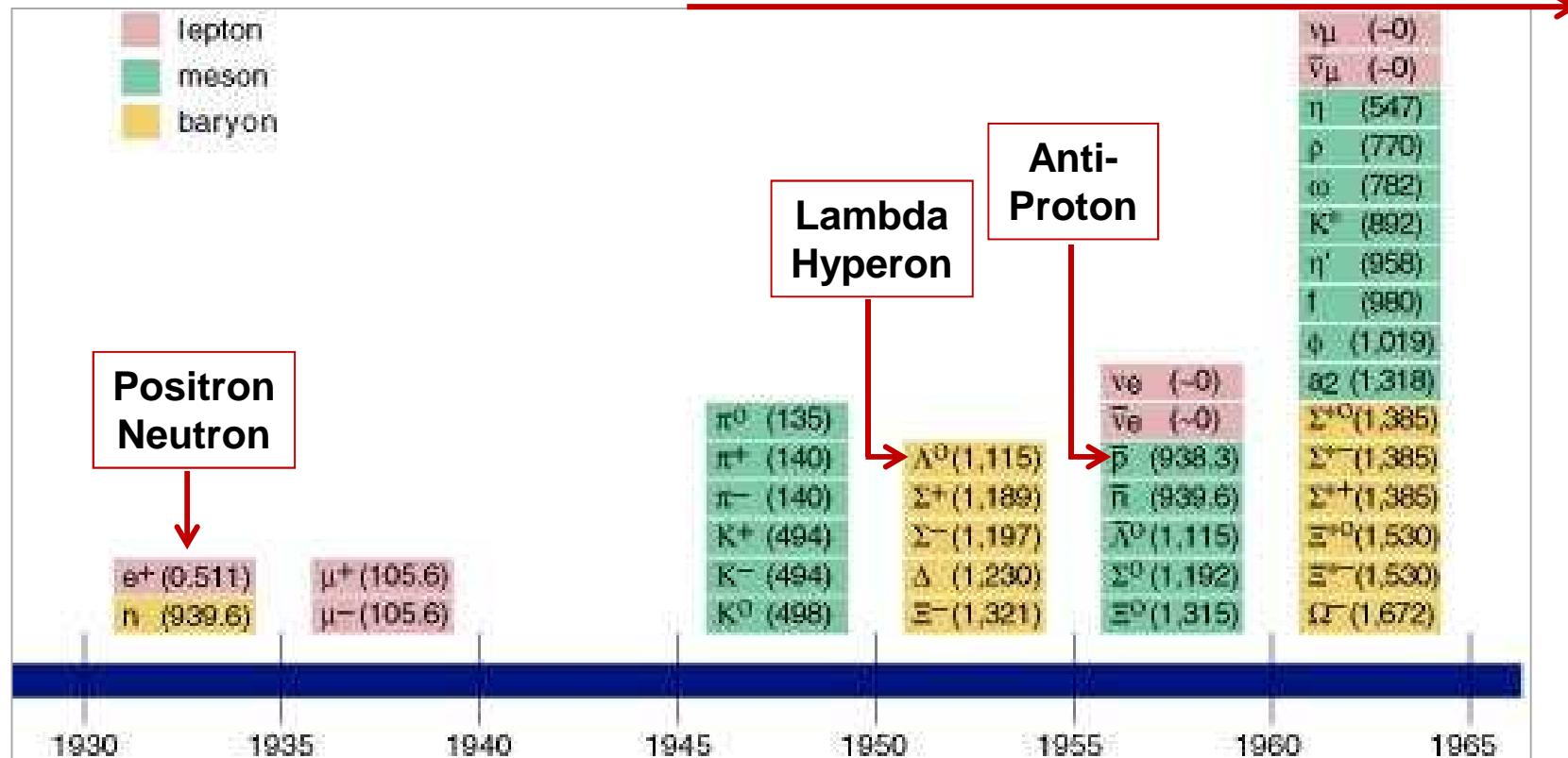
Introduction – History



Duplication: **matter** and **anti-matter**

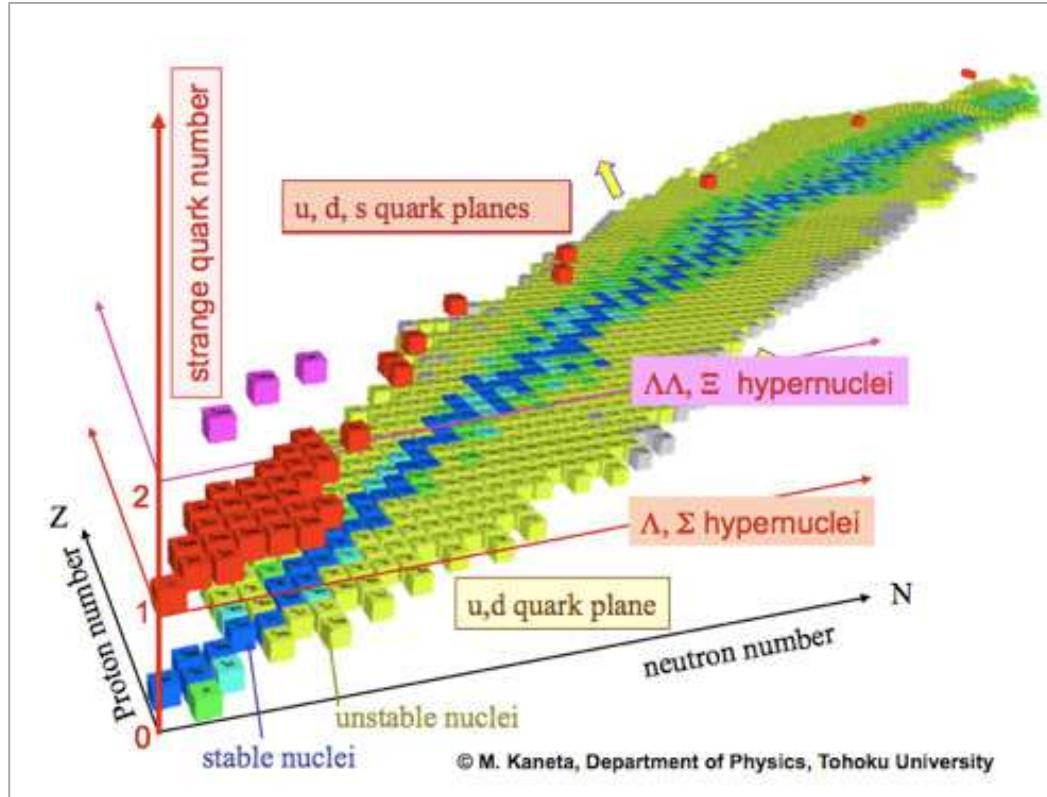
Introduction – History

The era of accelerators ...



Observation of a „zoo“ of sub-atomic particles and anti-particles

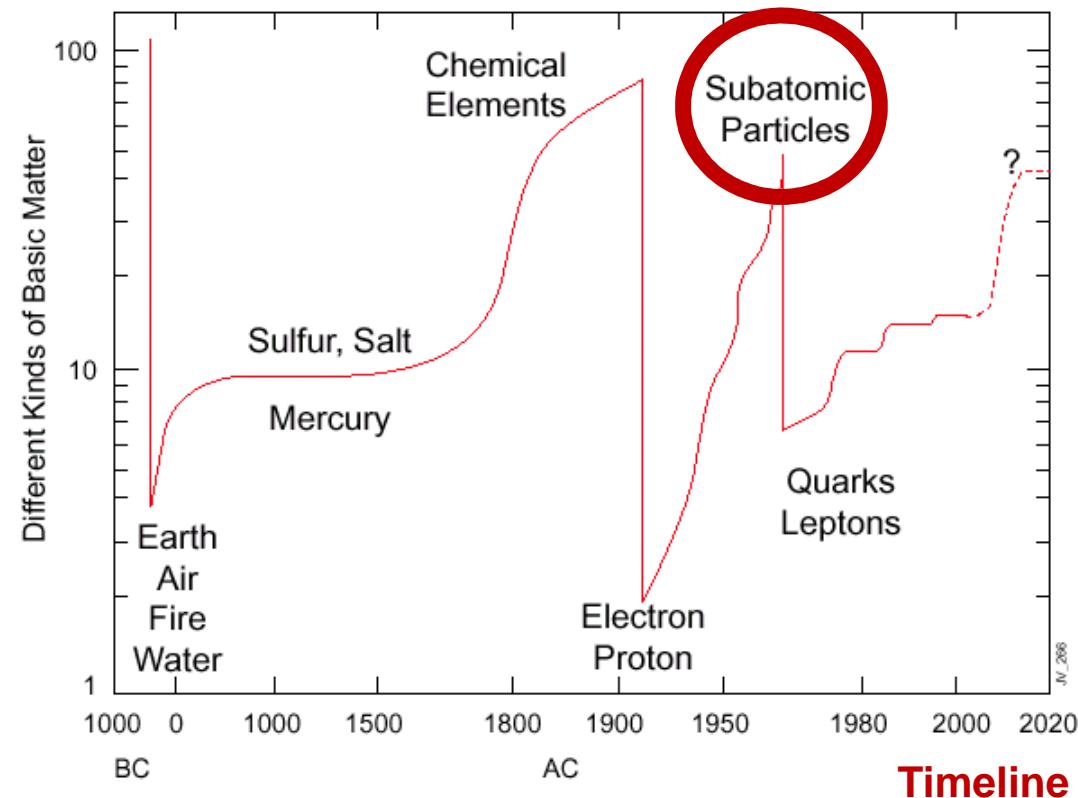
Introduction – History



“Hypernuclei”

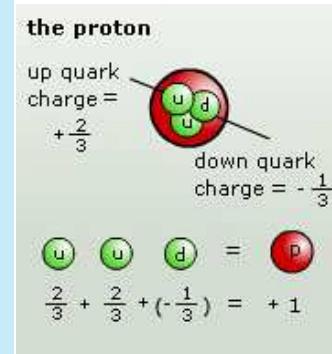
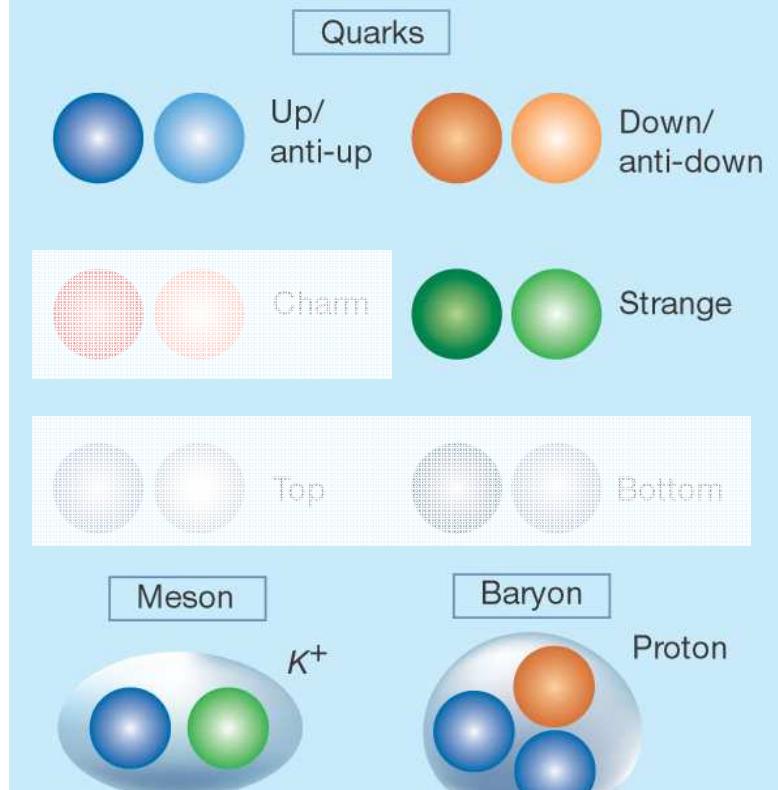
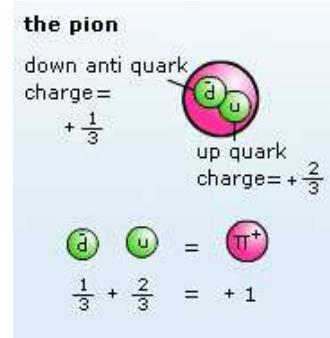
The extended „Table of Isotopes“

Introduction – History



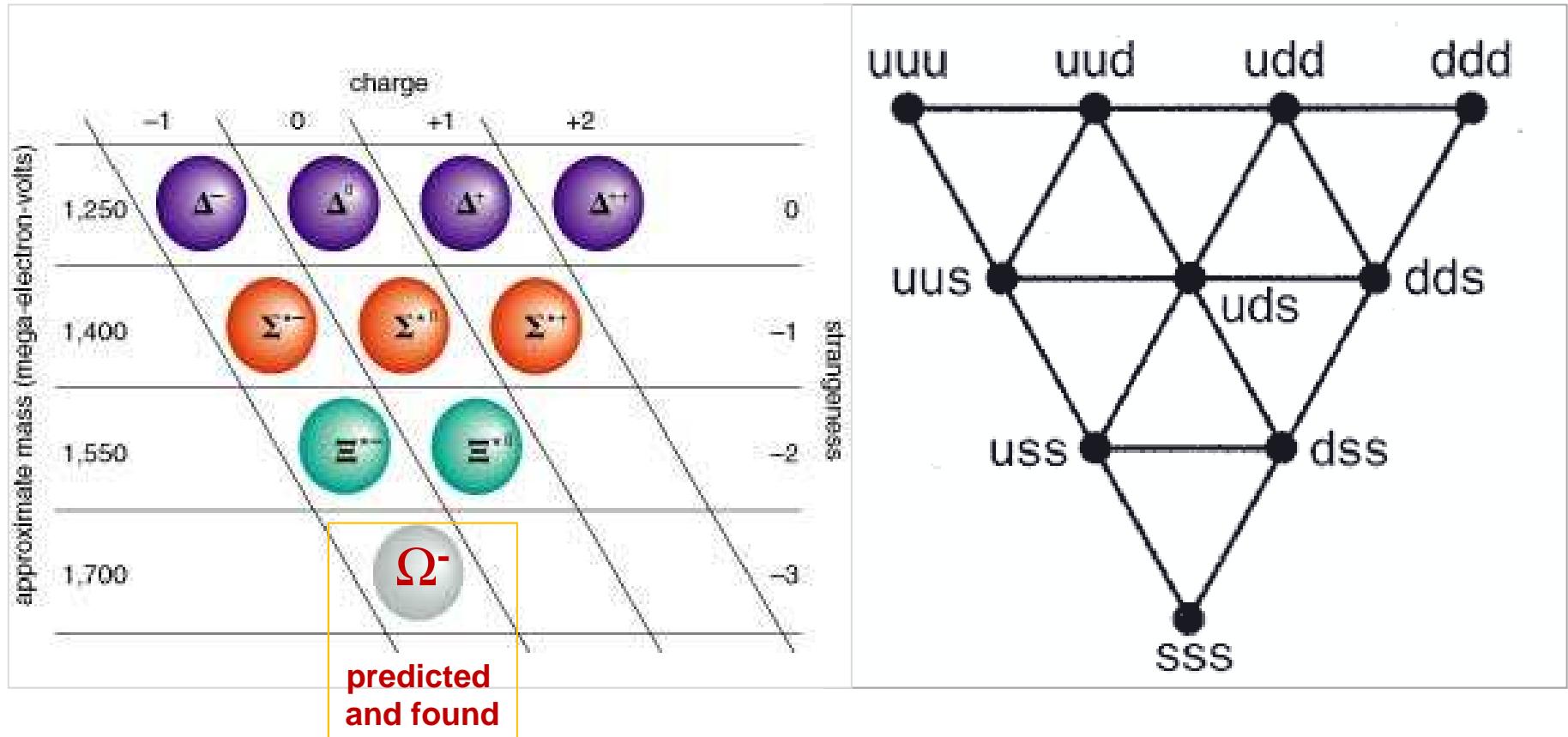
The up's and down's in the number of matter constituents

Introduction – History



M. Gell-Mann and G. Zweig (1964): **Quarks** („Aces“)

Introduction – History



Hadrons (baryons & mesons) appear in „multiplets“: all members found



გმადლობთ